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for ideal gas on a pv diagram an adiabatic curve is always steeper than an isotherm if  
they intersect the ideal gas equation  $pV^n = \text{constant}$  p is pressure v is volume n is number of  
moles r is ideal gas

the graph has a slope of  $\frac{3}{4}$  and the y intercept is at 0 3 the slope intercept form of the  
graph is  $y = \frac{3}{4}x + 3$  standard form  $3x - 4y + 12 = 0$  for the third problem given the table  
of

mar 10 2016 explanation  $y = mx + c$  is the equation of a line in slope intercept form where  
m represents the gradient slope and c the y intercept

equation is  $4y = x^4$  if slope is m and y intercept is c equation of line is  $y = mx + c$  and its x  
intercept is given by putting y 0 i e  $mx + c = 0$  or  $c = -mx$

the line perpendicular to the given line is  $y = \frac{2}{5}x + 3$  or  $y = 5x + 17$  you are given a point and  
an easy way to find slope so the most convenient equation is the point slope form of the  
equation of the

the way to solve this question is to first find the slope using the two points given and  
then use that slope like you would in point slope form the final equation  $y = 2x + 12$

problem 1 let  $x_1 = 1$   $y_1 = 3$   $x_2 = 2$   $y_2 = 9$  by slope formula  $m = \frac{y_2 - y_1}{x_2 - x_1} = \frac{9 - 3}{2 - 1} = 6$  by point  
slope form  $y - y_1 = m(x - x_1)$  we have the equation  $y - 3 = 6(x - 1)$  or  $y = 6x - 3$

apr 9 2017  $y = \frac{3}{2}x + 27$   $y = mx + b$  calculate the slope m from the given point values solve  
for b by using one of the point values and check your solution using the other point  
values a line can

As recognized, adventure  
as without difficulty as

experience about lesson,  
amusement, as with ease

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